
	FACULTY OF ENGINEERING COURSE SYLLABUS FORM	Doküman Kodu	MF.FR.003
		Yayın Tarihi	06.09.2024
		Revizyon No	0
		Revizyon Tarihi	0
		Gizlilik Sınıfı	Hizmet içi

NE201 - MATERIAL SCIENCE I				
Course Code	Course Name			Semester
NE 201	MATERIAL SCIENCE I			Fall <input checked="" type="checkbox"/> Spring <input type="checkbox"/> Summer <input type="checkbox"/>
Hours			Credit	ECTS
Theory	Practice	Lab	4	6
3	0	2		

Course Details	
Department	Nanotechnology Engineering
Course Language	English
Course Level	Undergraduate <input checked="" type="checkbox"/> Graduate <input type="checkbox"/>
Mode of Delivery	Face to Face <input checked="" type="checkbox"/> Online <input type="checkbox"/> Hybrid <input type="checkbox"/>
Course Type	Compulsory <input checked="" type="checkbox"/> Elective <input type="checkbox"/>
Course Objectives	<p>The objective of this course is to provide students with a fundamental understanding of materials science by exploring the interrelationships between structure, properties, processing, and performance. Through topics such as atomic structure, crystallography, mechanical behavior, phase diagrams, and microstructural evolution, students will gain the foundational knowledge necessary for material selection, processing decisions, and engineering design.</p>
Course Content	<p>This course introduces the fundamental principles of materials science and engineering, with an emphasis on the relationship between structure and properties of materials. Topics include atomic structure and bonding, crystallography, crystal defects, diffusion, mechanical behavior, strengthening mechanisms, failure analysis (fracture, fatigue, creep), phase diagrams, the iron-carbon system, microstructural development, and phase transformations.</p>
Course Method/ Techniques	Lecture <input checked="" type="checkbox"/> Question & Answer <input checked="" type="checkbox"/> Presentation <input checked="" type="checkbox"/> Discussion <input checked="" type="checkbox"/>
Prerequisites/ Corequisites	No
Work Placement(s)	No

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Textbook/References/Materials

Textbook: Callister, W.D., & Rethwisch, D.G. (2018). *Materials Science and Engineering: An Introduction* (10th ed.). Wiley.

References:

Smith, W.F., & Hashemi, J. (2010). *Foundations of Materials Science and Engineering* (5th ed.). McGraw-Hill.


Askeland, D.R., & Wright, W.J. (2015). *The Science and Engineering of Materials* (7th ed.). Cengage Learning.

Course Category

Mathematics and Basic Sciences	<input checked="" type="checkbox"/>		Education	<input type="checkbox"/>
Engineering	<input checked="" type="checkbox"/>		Science	<input checked="" type="checkbox"/>
Engineering Design	<input type="checkbox"/>		Health	<input type="checkbox"/>
Social Sciences	<input type="checkbox"/>		Profession	<input checked="" type="checkbox"/>

Weekly Schedule


No	Topics	Materials/Notes
1	Introduction to Materials Science: Overview of material types (metals, ceramics, polymers, composites); historical development; role of materials in engineering design. Atomic Structure and Bonding: Fundamentals of atomic structure; quantum numbers; types of bonding	Callister, Chapter 1,2
2	Crystal Structures and Crystallography: Point defects, dislocations, and grain boundaries; vacancy and interstitial atoms; significance of imperfections.	Callister, Chapter 3
3	Imperfections in Solids: Point defects, dislocations, and grain boundaries; vacancy and interstitial atoms; significance of imperfections.	Callister, Chapter 4
4	Diffusion Mechanisms: Fick's First and Second Laws; steady and non-steady state diffusion; factors affecting diffusion.	Callister, Chapter 5
5	Mechanical Properties of Materials: Stress-strain behavior; elastic and plastic deformation; modulus of elasticity; ductility; resilience and toughness.	Callister, Chapter 6
6	Dislocations and Strengthening Mechanism: Edge and screw dislocations; slip systems; mechanisms of strengthening (grain size reduction, solid solution strengthening, strain hardening).	Callister, Chapter 7
7	Failure I: Fracture Mechanisms: Ductile vs brittle fracture; fracture toughness; Griffith theory; stress concentration and flaw sensitivity.	Callister, Chapter 8
8	Midterm Exam	
9	Failure II: Fatigue and Creep: S-N curves; fatigue limit; fatigue crack growth; creep mechanisms; creep curve; rupture and steady-state creep.	Callister, Chapter 8
10	Introduction to Phase Diagrams: Phase rule; single-component systems; interpretation of phase diagrams; equilibrium phases.	Callister, Chapter 9
11	Binary Phase Diagrams: Binary isomorphous and eutectic systems; tie line and lever rule; microstructural development in alloys.	Callister, Chapter 9
12	Iron-Carbon Phase Diagram: Cementite and ferrite; eutectoid and eutectic reactions; pearlite, bainite, martensite transformations.	Callister, Chapter 9
13	Development of Microstructures: Effects of cooling rates; microstructure-	Callister, Chapter 9

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	property relationships; heat treatment concepts.	
14	Phase Transformations: Nucleation and growth; TTT and CCT diagrams; austenite decomposition; transformation kinetics.	Callister, Chapter 10
15	Review and Problem-Solving	Chapter 8-10
16	Final Exam	Chapter 1-10

Assessment Methods and Criteria		
In-term studies	Quantity	Percentage
Attendance		
Lab	4	15%
Practice		
Fieldwork		
Course-specific internship		
Quiz/Studio/Criticize	2	10%
Homework		
Presentation / Seminar		
Project		
Report		
Seminar		
Midterm Exam	1	25%
Final Exam	1	50%
	Total	100%
Contribution of Midterm Studies to Success Grade		50
Contribution of End of Semester Studies to Success Grade		50
	Total	100%

ECTS Allocated Based on Student Workload			
Activities	Quantity	Duration (Hrs)	Total Workload
Course Hours	14	3	42
Lab	4	2	8
Practice			
Fieldwork			
Course-specific Work Placement			
Out-of-class study time	14	3	42
Quiz/Studio/Criticize	2	7	14
Homework			
Presentation / Seminar			
Project			
Report			
Midterm Exam and Preparation for Midterm	1	20	20
Final Exam and Preparation for Final Exam	1	24	24
Total Workload			150
Total Workload / 25			150/25
ECTS Credit			6

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Course Learning Outcomes	
No	Outcome
L1	Understand the fundamental concepts of materials science, including atomic structure, bonding, and crystallography.
L2	Analyze material imperfections, diffusion mechanisms, and their effects on properties.
L3	Evaluate mechanical properties, strengthening mechanisms, and material failure types.
L4	Interpret phase diagrams, phase transformations, and microstructural evolution.
L5	Apply problem-solving skills to material selection and engineering applications

Contribution of Course Learning Outcomes to Program Competencies/Outcomes												
<i>Contribution Level: 1: Very Slight, 2: Slight, 3: Moderate, 4: Significant, 5: Very Significant</i>												
	P1	P2	P3	P4	P5	P6	P7	P8	P9	P10	P11	Total
L1	5	4	3	4	3	2	2	3	3	2	2	
L2	5	5	3	3	3	2	2	3	2	2	2	
L3	4	5	4	4	3	3	3	4	3	3	2	
L4	3	4	5	5	4	3	4	5	4	4	3	
L5	2	3	4	5	4	4	4	5	5	5	4	
Total												